This article was downloaded by:

On: 26 January 2011

Access details: Access Details: Free Access

Publisher Taylor & Francis

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-

41 Mortimer Street, London W1T 3JH, UK



Nucleosides, Nucleotides and Nucleic Acids

Publication details, including instructions for authors and subscription information: http://www.informaworld.com/smpp/title~content=t713597286

The Synthesis of Cyclonucleotides with Fixed Glycosidic Bond Linkages as Putative Agonists for P2-Purinergic Receptors

Girolamo Tusa^a; Juta K. Reed^a

^a Department of Chemistry, University of Toronto at Mississauga, Mississauga, Ontario

To cite this Article Tusa, Girolamo and Reed, Juta K.(2000) 'The Synthesis of Cyclonucleotides with Fixed Glycosidic Bond Linkages as Putative Agonists for P2-Purinergic Receptors', Nucleosides, Nucleotides and Nucleic Acids, 19: 4, 805 - 813

To link to this Article: DOI: 10.1080/15257770008035026

URL: http://dx.doi.org/10.1080/15257770008035026

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: http://www.informaworld.com/terms-and-conditions-of-access.pdf

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

THE SYNTHESIS OF CYCLONUCLEOTIDES WITH FIXED GLYCOSIDIC BOND LINKAGES AS PUTATIVE AGONISTS FOR P2-PURINERGIC RECEPTORS

Girolamo Tusa and Juta K. Reed*

Department of Chemistry, University of Toronto at Mississauga,
3359 Mississauga Rd., Mississauga, Ontario. L5L 1C6

Abstract: Cyclonucleotides with fixed glycosidic bond linkages were investigated as possible ligands for purinoceptors in PC12 cells. P2Y₂-purinoceptors were not activated by the ATP analogue, 8,2'-thioanhydroadenosine-5'-triphosphate (4) and only weakly by the UTP analogue, 2,2' -anhydrouridine-5'-triphosphate (6). However, both analogues were agonists for P2X₂-purinoceptors although the potencies were approximately 30-fold less than that of the parent nucleotides.

Adenine nucleotides are ubiquitous signaling molecules that target a variety of cell types through the interaction with a heterogeneous group of cell surface P2-purinoceptors. The physiological agonist for the P2-purinoceptor is usually ATP, although some subtypes can be activated by uridine nucleotides (UTP and UDP) or ADP. The P2-purinoceptors comprise two large superfamilies, the G-protein-coupled metabotropic receptors (P2Y_{1-n}) and the ligand-gated cation channels (P2X_{1-n})¹. Many cell types express multiple P2-purinoceptor subtypes and it would be of great advantage to selectively target individual receptor classes. To this end, many ATP analogues with substitutions on the base or polyphosphate tail have now been shown to display receptor specificity². Most of these are agonists with varying degrees of potencies; few receptor-specific antagonists have been identified to date.

Nucleotides, by virtue of free rotation around the glycosidic bond are conformationally flexible and can adopt either the *anti* or *syn* conformation. Nonetheless, one confomer may be preferred in solution or even within a ligand binding site in a protein^{3, 4}. ATP in solution prefers the *anti*-conformation⁵ with the plane of the base perpendicular and bisecting the sugar residue. In this paper we describe the synthesis of

^{*}To whom correspondence is addressed

two conformationally locked "cyclonucleotides" with fixed glycosidic bond linkages, 8,2'-thioanhydroadenosine-5'-triphosphate (4) and 2,2'-anhydrouridine-5'-triphosphate (6) (FIG.1). The purpose of this study was to attempt to determine if purinoceptors display conformation-based selectivity of the ligand. By using the preferred conformer of the nucleotide as a template, it may be possible to design more selective and potent ligands for these receptors.

The bioactivity of these two nucleotide analogues was determined in PC12 cells, a cell line that expresses multiple P2-purinoceptors⁶⁻⁸ that can be distinguished experimentally. The P2X₂-purinoceptor has been cloned from PC12 cells⁹, and P2Y₂-purinoceptors have been identified in PC12 cells by Northern analysis using a P2Y₂ clone¹⁰.

EXPERIMENTAL

Synthesis of cyclonucleotides: 8,2'-Thioanhydroadenosine-5'-triphosphate (4) was prepared by phosphorylation of the parent nucleoside, 8,2'-thioanhydroadenosine (3). The synthesis of 8,2'-thioanhydroadenosine (3) was based on a procedure described by Ogilvie et al. 11 with modifications for larger scale preparation which allowed all intermediates to be identified. The procedure involved four steps; bromination of adenosine to obtain 8-bromoadenosine (1), formation of the 2',3'-cyclic carbonate, 8bromoadenosine-2',3'-carbonate (2) which upon treatment with thiourea and refluxing, cyclizes to yield the sulfur-bridged nucleoside, 8,2'-thioanhydroadenosine (3). intermediates were obtained in high purity and characterized by ¹H-NMR, HPLC, FT-IR and FAB-MS. The 2', 3'-substitution in (2) was confirmed by a downfield shift in the ¹H-NMR signals (2'H, 6.26 ppm and 3'H, 5.60 ppm) compared to adenosine. The procedure for the formation of the triphosphate product was adapted from Zimmet et al. 12 The unprotected nucleoside (3) (1.1 mmole, 0.390 g) was reacted with POCl₃, (2.2 mmole, 205 µL) under anhydrous conditions in the presence of 1,8-bis(dimethylamino) naphthalene (proton sponge, 1.7 mmole, 0.234 g, 1.5 equiv. Aldrich) and trimethyphosphate (15 mL). After 2 hours at 0°C an excess (5 equiv.) of 0.5 M tri-nbutylammonium pyrophosphate in DMF was added quickly and allowed to react for 1 min. The reaction was quenched with 0.2 M aqueous triethylammonium bicarbonate (TEAB) (100 mL) and agitated gently for 30 min at room temperature to destroy metaphosphates. The selective phosphorylation of the 5'-hydroxyl was achieved by maintaining low temperatures. The nucleoside phosphate products were separated by DEAE-Sephadex-25 eluted with a linear gradient of TEAB (0-700 mM). The fractions containing the triphosphate were pooled and lyophilized to obtain 8,2'-thioanhydroATP (4) as the triethylammonium salt (translucent glass). The triethylammonium salt was

8,2'-ThioanhydroATP (4)

2,2'-AnhydroUTP (6)

FIG. 1.

dissolved in methanol (0.05 M) and five volumes of sodium perchlorate (15 equiv. in acetone) were added. The precipitated sodium salt was collected by centrifugation, washed with acetone (4 x 5 mL), and dried over P_2O_5 under vacuum. This afforded the sodium salt of the nucleoside-5'-triphosphate (4), (102.2 mg, 0.17 mmole, yield 15%): 1H -NMR (200 MHz, D_2O) δ 3.94 (m, 2H, 5'-H,5"-H, $J_{H5'/5",P}$ = 6.6 Hz), 4.39 (q, 1H, 4'-H, $J_{H4',5'/5"}$ = a 5.2 Hz), 4.71 (t, 1H, 3'-H, $J_{H3',4'}$ = 3.7 Hz), 4.97 (dd, 1H, 2'-H, $J_{H2',3'}$ = 3.0 Hz), 6.53 (d, 1H, 1'-H, $J_{H1',2'}$ = 7.4 Hz), 8.00 (s, 1H, 2-H adenine); ^{31}P -NMR (200 MHz, D_2O , pH 7) δ -22.48 (br s, 1P, βP), -10.87 (d, 1P, αP, $J_{\alpha\beta}$ = 20.2 Hz), -9.79 (d, 1P, γP, $J_{\gamma\beta}$ = 18.3 Hz); FAB-MS calculated for $C_{10}H_{10}O_{12}N_5P_3^{4-}$ of the free acid [M^{4-}] = 517.2, found 541.9 [M^{4-} + 2 H^+ + N_a^+]; HPLC: reverse-phase C18 column (4.6 x 250 mm), eluted isocratically for 40 min (0.1 M K_2 HPO₄, 0.2 mM (Bu)₄NH⁺SO₄ , 2.5% MeOH, pH 6.5, 1 ml/min) purity > 99%, retention = 33.6 min.

2,2'-Anhydrouridine-5'-triphosphate (6) was prepared by phosphorylation of 2,2'-anhydrouridine (5) (1.51 mmoles, 0.342 g) as outlined above. 2,2'-Anhydrouridine (5)

was prepared from uridine by treatment with diphenylcarbonate in DMF¹³. Since the cyclonucleoside (5) is susceptible to acid hydrolysis at the bridging oxygen, an excess of proton sponge (1.7 equiv.) was required to protect the nucleoside during phosphorylation. Identical work-up and DEAE-Sephadex-25 purification yielded the triethylammonium triphosphate salt, which was converted to the sodium salt, to afford 2,2'-anhydroUTP (6) as an off-white powder (148 mg, 0.27 mmoles, yield 18%): 1 H-NMR (200 MHz, D₂O) 1

^a Due to the near equivalence of the 5' and 5" protons only an average J_H is reported.

Biological assays: PC12 cells were cultured in RPMI media supplemented with 5% fetal calf serum, 5% heat-inactivated horse serum and 1% penicillin-streptomycin. P2Y₂ activity was determined by measuring the agonist-induced increase in intracellular Ca²⁺ using fura-2¹⁴ as described by Raha et al. 15. P2X₂ activity was monitored using the fluorescent potential-sensitive anionic dye, bisoxonol (bis[1,3-diethylthiobarbiturate]-trimethine oxonol) as outlined previously 6.

RESULTS AND DISCUSSION

The major conformational variations in nucleotides are associated with modes of puckering of the ribofuranose ring and with rotations around the glycosidic and the 4',5' bonds. Based on X-ray crystallography, computer simulations and NMR spectroscopy of nucleotide conformations,^{5, 17-18} it has been suggested that there is a low glycosidic rotational barrier with the *anti* rotomer in large excess over the *syn* rotomer.

Where the base and sugar are free to rotate, the *syn* and *anti* designations reflect a range of possible glycosidic bond angles as illustrated in FIG. 2 and a more accurate description of the glycosidic conformation requires defining the dihedral angle χ_{CN} . For purines, a χ_{CN} of 0° is when O(1') of the sugar and C(8) of the base are eclipsed¹⁹. For the pyrimidines, a χ_{CN} of 0° is when the O(1') of the sugar and C(6) of the base are eclipsed²⁰. From X-ray crystallography data of the cAMP analogue, 8,2'-thioanhydroadenosine-3',5'-cyclic monophosphate²¹, the adenine base is fixed with an χ_{CN} of 122° (high *anti* region) compared to the χ_{CN} for ATP of 39 and 69° 17 still in the designated anti range.

Several enzymes have been shown to require the nucleoside or nucleotide substrate in the favoured anti or high anti conformation. NMR studies on the

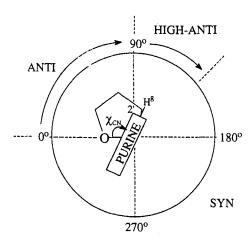


FIG. 2. Schematic representation of the position of the base in ATP. For 8,2'-thioanhydroATP the χ_{CN} is 122° (high-anti region).

conformation of ATP bound to rabbit muscle creatine kinase suggest that ATP binds preferentially in the anti conformation³. Adenosine deaminase favours adenosine analogues that possess high anti character as substrates⁴. Adenosine analogues favouring the high anti conformation have been shown to activate adenosine receptors in lymphocytes and cardiac cells²². With these studies in mind, the conformational specificity of the P2-purinoceptors in PC12 cells was assessed using 8,2'thioanhydroATP (4), an anti-locked rotomer of ATP and 2,2'-anhydroUTP (6) a synlocked rotomer of UTP. Although syn rotomers are not highly favoured in solution there are examples of enzymes preferring syn. For example, potent inhibitors of uridine phosphorylase, such as 2,2'-anhydro-5-substituted uridines, are fixed rotomers of uridine with high syn character²³. The X-ray crystal structure of an ATP boundphosphoenolpyruvate carboxykinase (PCK) complex stabilizes a syn conformation of the base²⁴. An investigation of the energy profiles of the syn and anti conformers of nucleosides by quantum mechanical methods suggests the difference in energy between syn and anti conformers is less than 1 kcal-mole-122 Thus the small energetic cost incurred by binding ATP when it is in a syn conformation appears to be outweighed by the favourable interactions it promotes with the enzyme.

Activation of P2Y₂-purinoceptors in PC12 cells elicits a transient increase in $[{\rm Ca}^{2^+}]_i$ triggered by intracellular release from the endoplasmic reticulum as well as from ${\rm Ca}^{2^+}$ influx. FIG. 3 shows a series of typical calcium transients; ATP and UTP at concentrations as low as 10 μ M produced appreciable responses, while the addition of

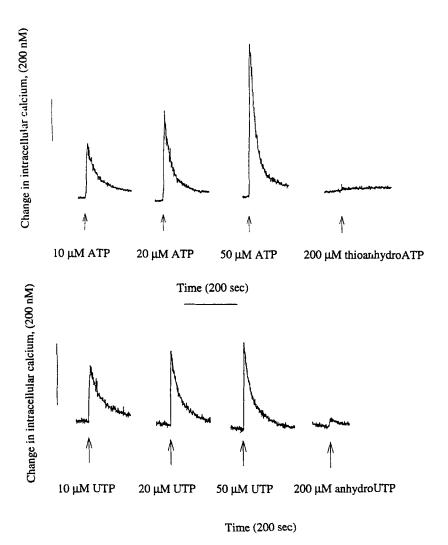


FIG. 3. Activation of P2Y₂-purinoceptors in PC12 cells leads to a transient increase in intracellular calcium as measured using fura-2.

200 μ M 8,2'-thioanhydroATP (4) was without effect. In separate experiments, concentrations as high as 500 μ M also produced no response (data not shown). 200 μ M 2,2'-anhydroUTP (6) only produced a greatly reduced although reproducible transient. Neither analogue (200 μ M) antagonized the response elicited by ATP or UTP (20 μ M) (data not shown).

Activation of $P2X_2$ -purinoceptors leads to membrane depolarization and in intact PC12 cells this can be measured using potential sensitive lipophilic anionic dyes such as

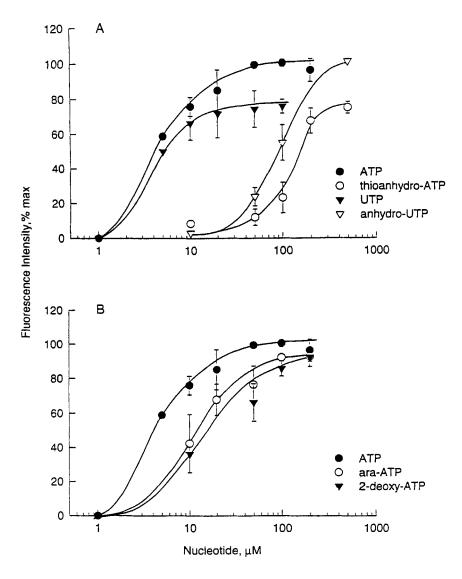


FIG. 4. Agonist-induced depolarization of PC12 cells measured by an increase in bisoxonol fluorescence (value ± SEM)

bisoxonol. FIG. 4 shows the dose response profiles for ATP, UTP, the locked nucleotide analogues and the two 2'-modified nucleotides araATP and 2'-deoxyATP. Both 8,2'-thioanhydroATP (4) (EC $_{50}$ ~150 μ M) and 2,2'-anhydroUTP (6) (EC $_{50}$ ~100 μ M) are agonists for P2X $_2$ -purinoceptors in PC12 cells although both are significantly weaker than the parent nucleotides (EC50 ~4 μ M). Both araATP and 2'-deoxyATP were clearly good agonists (EC $_{50}$ ~12-15 μ M) suggesting that the reduced efficacy of the

conformationally locked analogues is not due to loss of the 2'-hydroxyl and possible H-bonding capacity nor can it be attributed solely to a loss of stereochemistry at that position.

The lack of P2Y₂-purinoceptor agonist activity of either locked nucleotide suggests that the ligand recognition site in these receptors may not accommodate the molecule in either the constrained high *anti* or even a smaller but inflexible *syn* conformation. Alternatively, the constraints imposed by either the sulfur or oxygen bridge between the base and C(2') may distort the molecule especially the sugar pucker. The observation that both molecules are significantly more potent with P2X₂-purinoceptors may not be altogether surprising given that the membrane topography of these receptors is completely different from that of the metabotropic receptors. Indeed it would appear that the ligand recognition site in P2X₂-purinoceptors may accommodate far more base flexibility than that of the P2Y₂-purinoceptor. Whether this selectivity holds for other members of the P2 superfamilies remains to be determined.

REFERENCES

- 1. Burnstock, G. Neuropharmacol., 1997 36, 1127-1139.
- 2. Jacobson, K. A.; Jarvis, M. F. in "Purinergic Approaches in Experimental Therapeutics", K. A. Jacobson, M. F. Jarvis, Ed., Wiley-Liss Press, 1997, pp.101-128.
- Rosevear, P. R.; Powers, V. M.; Dowhan, D.; Mildvan, A. S.; Kenyon, G. L. Biochemistry, 1987 26, 5338-5344.
- 4. Hampton, A.; Harper, P. J.; Sasaki, T. *Biochemistry*, **1972** 11, 4736-4739.
- 5. Nair, V.; Young, D. A. Magn. Res. Chem., 1987 25, 937-940.
- 6. de Souza, L. R.; Moore, H., Raha, S.; Reed, J. K. J. Neurosci. Res., 1995 41, 753-763.
- 7. Nikodijevic, B.; Sei, Y.; Shin, Y.; Daly, J. W. Cell Mol. Neurobiol., 1994 14, 27-47.
- 8. Barry, V. A.; Cheek, T. R. J. Cel. Sci., 1994 107, 451-462.
- 9. Brake, A. J.; Wagerbach, M. J.; Julius, D. Nature, 1994 371, 519-523.
- 10. Raha, S.; Reed, J. K. (unpublished)
- 11. Ogilvie, K. K.; Slotin, L.; Westmore, J. B.; Lin, D. Can. J. Chem., 1972 50, 2249.
- 12. Zimmet, J.; Järlebark, L.; Hammerberg T.; Van Galen P.; Jacobson, K.; Heilbronn, E. *Nucleosides and Nucleotides*, **1993** 12, 1-20.
- 13. Hampton, A.; Nichol, A. W. Biochemistry, 1966 5, 2076-2081.
- 14. Grynkiewicz, G.; Poenie, M.; Tsien, R. Y. J. Biol. Chem., 1985 260, 3440-3450.
- 15. Raha, S.; de Souza, L. R.; Reed, J. K. J. Cellular Physiol., 1993 154, 623-630.
- Rink, T. J.; Montecucco, C.; Hesketh, T. R.; Tsien, R. Y. Biochim. Biophys. Acta, 1980 595, 15-30.
- 17. Sundaralingam, M. Ann. N. Y. Acad. Sci., 1975 255, 3-43.

- 18. Viswanadhan, V. N.; Ghose A. K.; Weinstein, J. N. *Biochim. Biophys. Acta*, **1990** 1039, 356-366.
- 19. Paul, M. L.; Miles, D. L.; Cook, M. A. J. Pharmacol. and Exp. Therap., 1982 222, 241-245.
- 20. Hatano, M.; Kaito, A.; Tajiri, A.; Ueda, T.; Shibuya, S. *Nucleic Acid Res. Sym. Ser.* 8, **1980**, 125-128.
- 21. Tomita, K.; Tanaka, T.; Yoneda, M.; Fujiwara, T.; Ikehara, M. Acta Crystallogr. Sect. A, 1972 28, S45.
- 22. Miles, D. L.; Miles, D. W.; Eyring, H. Proc. Natl. Acad. Sci., 1977 74, 2194-2198.
- 23. el Kouni, M. H.; Naguib, F. N. M.; Panzica, R. P.; Otter, B. A.; Chu. S. H.; Gosselin, G.; Chu, C. K.; Schinazi, R. F.; Shealy, F.; Goudgaon, N.; Ozerov, A. A.; Ueda, T.; Iltzsch, M. H. Biochem. Pharm., 1996 51, 1687-1700.
- 24. Tari, L. W.; Matte, A.; Pugazhenthi, U.; Goldie, H.; Delbaere, L. T. *Nat. Struct. Biol.* **1996** 3, 355-363.